



## Molecular Modelling Applied for Carbon Nano Materials

Hend A. Ezzat<sup>a</sup>, Medhat A. Ibrahim<sup>b,\*</sup>, Hanan El-Haes<sup>c</sup>



CrossMark

<sup>a</sup>Nano Technology Unit, Solar and Space Research Department, National Research Institute of Astronomy and Geophysics (Nano NRIAG), 11421 Helwan, Cairo, Egypt

<sup>b</sup>Molecular Spectroscopy and Modelling Unit, Spectroscopy Department, National Research Centre, 33 El-Bohouth St., 12622, Dokki, Giza, Egypt

<sup>c</sup>Physics Department, Faculty of Women for Arts, Science and Education, Ain Shams University, 11757 Cairo, Egypt

### Abstract

Molecular modelling is applying computer software to describe the molecular systems. This leads to understating many systems and structures in chemical, biological systems. Molecular modelling is now widely used in much basic as well as applied science. On the other hand, carbonaceous materials which also known as carbon nano materials have attracted interests of many researchers according to their amazing special structures and extraordinary electronic properties. So that, research on the carbon nano materials are now increasing rapidly. Accordingly, carbon nano materials are surveyed with special care to fullerene, carbon nanotubes and graphene as well as their based systems. The review include how can molecular modelling describe the physical, chemical and functionality of the carbon nano materials. The review includes the following points

1. Introduction
2. Molecular modelling
3. Calculated parameters through molecular modelling
4. Carbon nano materials
5. Fullerene based systems
6. CNT based systems
7. Graphene based systems
8. Modelling other forms of carbon
9. Conclusion and outlook

**Keywords:** Molecular modelling ; fullerene; CNT; graphene..

### 1. Introduction

Nanotechnology is worldwide science and technology find its applications in all fields, in all fields one can find nanoscale, nanoparticle,

nanophase, nanocrystal, or nanomachine. So that, this field attracts worldwide attention. Simply nano scale materials are those with dimension in nano meters which is length scale, in this sense 1 nm is equal to a billionth of a meter ( $10^{-9}$  m) [1].

\*Corresponding author e-mail: [medahmed6@yahoo.com](mailto:medahmed6@yahoo.com); (Medhat A. Ibrahim).

Receive Date: 29 March 2020, Revise Date: 19 April 2020, Accept Date: 12 July 2020

DOI: 10.21608/EJCHEM.2020.26861.2551

©2020 National Information and Documentation Center (NIDOC)

This class of materials is something between chemical interactions which took place among atoms and within molecules in range below 1 nm and condensed matter physics which is representing clusters of materials gathered in crystals contains huge numbers of atoms. This makes several scientists ranging materials between 1 nm to 100 nm not within chemistry or solid-state Physics. This in turn requires new concepts and design of equipment's to follow up this new branch also requires approximations of the theoretical methods to follow up changes in this nano scale materials. Better understanding of this range leads to continuous developments and achievements in many areas whereas nano materials are applied. It could be here stated that, nanotechnology is the manipulation of matter with at least one dimension sized from 1 to 100 nanometers [2,3].

It is worth to mention that for manipulation of nano scale materials molecular modelling is a promising field. Moreover, carbon nano materials are an important class of nano materials due to its novelty applications in many areas. Accordingly, an introduction to molecular modelling is presented then carbon nano materials will be survived with the focus to their possible studies with molecular modelling .

## 2. Molecular modelling:

Molecular modelling is simulation conducted for molecular systems to understand the molecular behaviour. It is a class of computational work based on the quantum mechanics designed to study the chemical structures. It is an effective tool in materials science, physics and chemistry. It could be applied whereas experimental facilities are limited or unavailable or ethically not allowed for many systems such as biological systems [4-9]. It computes the energy of a particular molecular system, which leads to predict geometrical parameters; thermochemical parameters and vibrational frequencies including Infrared and Raman beside many other physical as well as chemical important parameters.

Such class of computational work pointed toward enhance the communication between experimental and theoretical research on both existing and new advanced findings based on their amazing applications. It is now worldwide applied for many systems and molecules covering many areas of both basic and applied science [10-14].

Recent applications of molecular modelling are guiding researchers to elucidate the molecular structure and chemical interactions of molecules in many areas of applied research [15-17].

Molecular modelling consists of molecular mechanics and electronic structure method [18], both could be summarized as in the following:

**Molecular mechanics:** It applies the laws of classical physics to predict the structures and the properties of the molecules. It performs computations based upon the interaction among the nuclei. Electronic effects are approximated, this makes the computations quite inexpensive, and used for very large system.

**Electronic structure method:** It applies schroedinger wave equation. Practically exact solution of schroedinger equation is not enough so, electronic structure method has many approximations to its solution. It has the following classes

*Semi-empirical methods*

*Ab initio*

*Density functional methods.*

More details about the basic principles of such classes of electronic structure methods were reported [18-20].

## 3. Calculated parameters through molecular modelling :

For electronic structure method and from theoretical point of view the model is an approximation to solve schroedinger wave equation. So, the model is a theoretical method with basis set.

As stated earlier [21] the method could be Ab initio or density functional theory. While the basis set is a mathematical representation of the orbitals. The combination between a theoretical procedure, and a basis set is used to approximate a solution for schroedinger wave equation. As it is a second order equation, it has two solutions. One of the most important parameters, which obtained through the first derivative, is the optimized geometry of the studied structure that predict bond lengths and bond angles of the structure. It localizes the lowest energy molecular structure in close proximity to the specified starting structure. It depends primarily on the gradient of the energy, i.e. it is the first derivative of energy with respect to atomic position. Moreover, through first derivative one can also obtain the total energy and total dipole moment. Otherwise, the second derivative one can obtain

many parameters such as; vibrational frequencies including Raman and Infrared; polarizability; thermochemical parameters. It is worth to mention that these methods or the solutions could be compared with experimental results after so called corrections or scaling. As these methods contain systemic errors which could be corrected by so called scale factor. Only scaled data could be compared with experimental results.

### Carbon nano materials:

Carbon-based materials are now widely used for many applications such family include graphite, activated carbon, fullerene, carbon nanotubes, mesoporous carbon, diamond and recently graphene [21].

Figure 1 presents some types of carbon-based materials as graphite, fullerene, Single walled carbon nanotubes, Multiwalled carbon nanotubes, graphene and Diamond respectively.

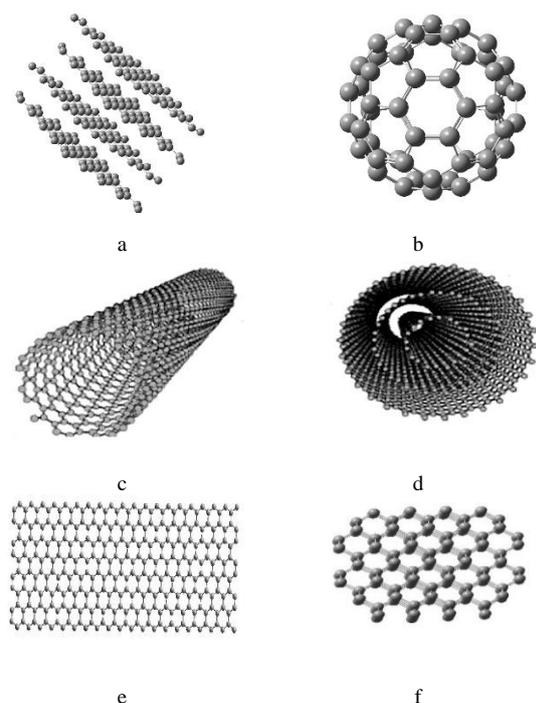


Fig. 1. Carbon based materials a- graphite, b-fullerene, c- Single walled carbon nanotubes, d- Multiwalled carbon nanotubes, e graphene and f- Diamond.

In the following not all but only some members of carbon materials will be surveyed. Starting with fullerene, which is also termed as  $C_{60}$  is a member of carbon nano materials since its discovery is considered among the most attractive point of research [22]. Physically, it is described with

unusual magnetic properties which may be correlated to its icosahedral nature. It is corresponding to  $I_h$  symmetry, in addition, its magnetic susceptibility arises from the existence of  $\pi$ -electrons ring currents in its carbon spheroid. In 1991, Sumio Iijima discovered a byproduct of fullerene called carbon nanotubes (CNTs) of two types' single-walled nanotubes and multi-walled nanotubes. Single-walled nanotubes with growing diameters being arranged (like "Russian doll") in a concentric manner, while multi-walled nanotubes may consist of one rolled up graphene sheet [23]. Carbon nanotubes could be produced in considerable amount using catalytic decomposition of acetylene in the presence of supported Co and Fe catalysts [24]. Another method could be achieved by a 60 keV  $Ar^+$  ion bombardment with normal incident angle under high vacuum. In such method fullerene was first transformed into amorphous carbon then formed carbon nanotubes [25]. Both fullerene and Carbon nanotubes could be also produced using the conventional catalytic Chemical Vapor Deposition (CVD) with certain care single-walled carbon nanotubes could be also produced [26]. Other modifications for producing could be reported elsewhere [27-29]. Since the discovery of CNTs till now, it attracts interest due to their wide range of applications, including high strength [30], extraordinary flexibility [31], excellent electrical conductivity [32] and field emission properties [33], which promise tremendous applications in electron field emitter of displays [34], nanoscale electronic devices [35], biosensors [36], hydrogen storage [37] and fuel cell electrodes [38]. Graphene is a two-dimensional structure; its carbon atoms is considered as surface atoms [39]. This makes its electronic properties is changed with introducing atoms like transition metals and/or metal oxides [40-41] this could be also achieved with molecules as well [42]. When graphene is interacted with metal nanoparticles (Ag, Au, Pt and Pd) there is a significant charge-transfer interaction which dedicates this composite for many applications depending on their surface [43]. Although graphene is a member of carbon nano materials, it is now the parent of a new family of graphene-based materials [44]. Such a new family is not only for gas sorption but also for energy storage [45]. Increasing the applications of modified graphene comes from the fact that graphene properties are not only a function of its number of layers but also a function of the

structural defects [46]. As indicated earlier, such defects could be achieved with doping or decoration, which enhances the ability of graphene to carry out its task. Moreover, it becomes highly sensitive and selective to act as gas sensor. Continuous work on graphene-based materials indicated its suitability to act as electrochemical biosensors for different materials including ascorbic acid; dopamine; uric acid; amino acid tryptophan as well as detecting nitrite in human serum [47].

#### 4. Modelling carbon nano materials :

Applying molecular modelling could be effective tool to investigate different properties of carbon nano materials. In this section it will be directed to three members of the family namely fullerene, carbon nanotubes and graphene.

#### 5. Fullerene Based Systems:

As mentioned earlier fullerene is belonging to structures of ambiguous aromatic character; traditional measurements are not providing proper classification [48]. This in turn paves the way toward new powerful characterizing tools to investigate it. Doping and/or substitution could enhance the electronic properties but following up the effect of that could be described on the theoretical basis. Molecular modelling with Monte Carlo simulation level is utilized to calculate the Pauli paramagnetic susceptibility of  $A_3C_{60}$  ( $A=K, Rb$ ) compounds [49]. This confirms the findings that,  $C_{60}$  is an aromatic molecule with a vanishingly small ring current magnetic susceptibility [50]. Molecular modelling with different level of theories show the ability for calculating important physical and chemical parameters necessary for understanding the properties of fullerene. Quantum mechanical calculations using Ab initio was used to elucidate, the structural and electronic properties of small silicon clusters and endohedral metallofullerenes [51]. Some efforts were also utilized with Density Functional Theory, DFT for reporting the structural parameters then the stability of  $C_{60}CH_2$  [52]. Another level of theory at Ab initio was also conducted for studying stability of  $MC_{60}$  where M is Sc, Y, and K respectively [53]. The same level of theory was also consulted for describing in details the equilibrium structure of giant fullerenes [54]. Calculations upon inorganic fullerene spheroids were performed at semiempirical molecular orbital calculation. Some

important parameters were calculated including geometrical parameters, electronic properties, and then vibrational characteristics [55]. The fullerene family include another members and derivatives, the structure, stability and polymerization of  $C_{28}$  was calculated with ab initio quantum mechanical level [56]. Computational levels are also modified in order to follow up the changes in the  $C_{60}$  systems. The first-principles DFT calculations were utilized to describe the adsorption of  $C_{60}$  on Si (111) [57]. Another computational effort was carried out at Ab initio level to investigate the interaction between  $C_{60}$  and Si (100) [58]. Calculations are predicting the stability of other fullerene members so that, it is proven that, g- $C_{80}$  and g- $C_{240}$  cages are less stable and have smaller HOMO-LUMO gaps as compared with their graphite isomers [59]. Time-dependent DFT combined with sum-over-states method were utilized to estimate the static third-order optical susceptibility  $\chi(3)$  for BN fullerene materials [60]. Semiempirical calculations were proposed to study the structure and vibrational properties of  $C_{60}$ ,  $C_{80}$  as well as their epoxides [61]. As an application of the effect of doping, it is stated that  $C_{60}$  behave like superconductor when it is doped with K [62]. But this requires some kind of cooling around 18k. This paves the way toward  $C_{60}$  doping with alkali metals then superconducting properties achieved, including high critical magnetic fields [63]. Further efforts are then emphasis that, the alkali metal doped  $C_{60}$  are good candidates for superconductivity based on their unique electronic structure [64]. Based on Ab initio calculations, the phonon spectrum of  $K_6C_{60}$  are presented. The effects of doping upon the infrared frequencies and their intensities are identified and correlated with their physical origin. The results are discussed in detail for optically allowed modes [65]. Other efforts were carried out for doping, as a hole doped  $C_{60}$  at relatively higher temperature around 52 k [66]. Rather than doping with metals other way of interactions could be achieved with functional groups and/or other chemical organic structures. So that,  $C_{60}$  were interacted with  $CHCl_3$  and  $CHBr_3$  it is reported that an expansion in the lattice took place [67]. According the unique electronic properties of modified fullerene it is also applied as device. It is tested as photovoltaic cells when is prepared in polymeric matrix [68]. Experimental efforts are conducted to prepare filamentary K- $C_{60}$  superconductor by the suspension spinning method.

Results show superconductivity with  $T_c=18$  K, which was indicated by SQUID measurement [69]. This in turn enhances the applications of fullerene composite as polymer photovoltaic cells. Accordingly, conjugated polymer/  $C_{60}$  composite was prepared for this purpose [70].

$C_{60}$  was further mixed with polystyrene then, bistability in single layer devices was observed. This was the first principle toward the applications of such class of materials as devices in disposable printable electronics [71]. According to the amazing properties and applications of composite materials based on  $C_{60}$  it is combined with carbon nanotube for many advanced applications including X-ray, neutron as well as high-energy particle physics [72].

### 6. CNT based systems:

The structure of CNTs show abundant pores with large surface-to-volume ratios, this in turn enhances the process of adsorption/desorption of gases onto the CNTs surface [73-74]. Such process could change the physical properties of the CNTs surface and paves the way toward applications of CNTs in the field of gas sensors [75-76]. It is stated that, sensor based on CNTs and/or their derivatives are characterized by faster response as compared with traditional sensors. It is stated that, significant variations in the electronic properties of the CNTs is recorded when it acts as gas sensor. Accordingly, DFT calculations based ATK-VNL and Gaussian approach has been used to verify the sensing phenomena of CNTs and used effectively to follow the changes in the electronic properties. A case study is reported for  $H_2S$  sensing for pristine and functionalized zigzag [77]. The gas molecules that adsorb on the surface of CNTs, change the shape of CNTs and trigger redistribution of electrons, leading to a macroscopic change in their resistance. Batch experiments was confirmed by DFT calculations in order to conduct quantitative correlation between structural parameters and CNTs adsorption performance [78]. Molecular modelling analyses of the electronic properties of the CNTs leading the research towered further applications of CNTs. Based on high response, selectivity, high surface area it is reported that, CNTs are excellent candidate for different applications in many areas such as environmental monitoring, space, biomedical and pharmaceutical applications [79-88].

DFT is also confirming some experimental findings for the suitability of CNTs in biological applications. In this sense, beside experimental work, a detailed study on the interaction between pyrimethamine anticancer drug and (6, 0) zigzag single-walled carbon nanotube was performed by DFT/B3LYP and DFT/M06-2X with 6-31G\* level of theories [89].

### 7. Graphene based systems

DFT calculations were carried out for both graphene as well as their complexes to study their abilities for adsorptions of some ions. Coronene was used as graphene model system, complexation was described as exothermic and spontaneous in most cases. The spectral analysis indicated significant variation in electronic properties based upon complexation [90]. The first-principles DFT calculations was used to investigate the mechanism of oxygen reduction reaction in fuel cells. For such reason, copper-nitrogen embedded graphene ( $CuN_3$ -Gra) is introduced as an efficient electrocatalyst [91]. Molecular modelling analyses dedicate graphene as a catalyst according to its amazing properties including its like large surface area, high thermal and electronic conductivity, high mechanical strength and excellent chemical stability [92]. It is stated that, graphene is almost inert regarding the process of adsorption owing to the in-plane  $\pi$ -conjugation. It is suggested to enhance the surface reactivity of graphene, this is could be conducted as one replacing one or more of carbon atoms with heteroatoms, this could dramatically change the electronic properties [93-94].

As well as other members of carbon nano materials graphene show the potential applications as gas sensor. It has high specific surface area, extremely low Johnson noise, unusual carrier density dependent electrical conductivity and limited crystal defects [95-100]. Molecular modelling data dedicate graphene for novel and unusual applications. It could be applied as a heavy metal detector [101]. While it is reported that such applications could be carried out with graphene quantum dots [102]. Molecular modelling suggested that, the interaction of graphene with heavy metals leads to variation in electronic properties in terms of charge transfer and Schottky barrier height which leads to the change in the current flowing through the barrier.

Regarding the fullerene, CNTs and graphene it is clear that the investigation of electronic properties is important step toward understating the mechanism of interaction of such structures with their surrounding molecules, then it is important to understand the electronic properties for functionality and further applications of carbon nano materials. It is stated that molecular modelling methods specially those based on DFT are effective methods to investigate electronic properties of carbon nano materials [103-106].

### 8. Modelling other forms of carbon:

Exfoliated graphite (EG) is a promising material for many applications such as flow field plates for fuel cells, EMI shielding, vibration damping and stress and chemical sensing [107-108]. Owing to these wide range of application EG is exposed to molecular modelling to assess its different properties including the mechanical, thermal and electrical properties respectively [109].

Modelling was supporting experimental finding in order to optimize the application of graphite as electrodes with different thicknesses and porosities for high-energy-density Li-Ion batteries [110]. Diamond D5 substructures was subjected to molecular dynamics simulations [111]. The structural stability of such intermediates/fragments appearing in the construction/destruction of D5 net was investigated. The nanotwinned diamond films under nanoindentation was subjected to molecular dynamics simulations [112].

The rational design of carbon fibers with desired properties requires quantitative relationships between some parameters such as microstructure and resulting properties. Molecular modelling with different levels shows potential applications for predicting the microstructure evolution during the processes of carbonization which in turn is effective tools for tailoring the desired carbon fibers [113-

114]. It is also reported that, molecular modelling could be also utilized to study the mechanical behavior of carbon fiber-amine functionalized multiwall carbon nanotube/epoxy composites [115].

### 9. Conclusion and outlook:

Based upon the above considerations carbon nano materials have unique surface, physical as well as chemical properties which leads to amazing applications. Based on molecular modelling it is clear that, hetero atoms could dramatically alter the electronic properties which leads to further applications covering many areas of science and technology. It is stated that some important and simple parameters such as total dipole moment, band gap energy and molecular electrostatic potential could be important to understand the functionality of carbon nano materials [116-118].

It is now well known to utilize quantum mechanical methods for tailoring materials with special functions to act for certain applications. For examples the discovery of two-dimensional (2D) materials such as graphene, silicene, molybdenum disulfide, black phosphorus, and graphitic carbon nitride have received tremendous attention owing to their exceptional features with respect to quantum transport, photoelectric activity, and photocatalysis [119-120]. There are systematic errors within these molecular modelling methods which could be corrected with scale factor. For methods like DFT: B3LYP [121-123] the accuracy is comparable with experimental results after scaling the calculated results. So that, molecular modelling with different levels and routes are now ready to design new materials for future purposes. Finally, this review indicated that molecular modelling is a useful tool for studying carbon based materials as well as their derivatives. It is also of concern for many other systems and structures whereas the experimental tools are limited and/or unavailable [124-130].

### 10. References

- [1] Kenneth J. K., Nanoscale Materials in Chemistry, Wiley-Interscience, New York (2001).
- [2] Drexler, Eric K., Engines of Creation: The Coming Era of Nanotechnology. Doubleday (1986).
- [3] Drexler, Eric K., Nanosystems: Molecular Machinery, Manufacturing, and Computation. New York: John Wiley & Sons (1992).
- [4] Ali G. W., Abdel-Fattah W. I., Elhaes H., Ibrahim M. A., Spectroscopic and modelling analyses of bimolecular structure of corn silk, *Biointerface Research in Applied Chemistry*. 9 (6), 4481-4485 (2019).
- [5] Bayoumy A. M., Badry R., Gaber H. A., Elbiomy S. A., El Gabaly S. G., Abd ElAziz M. S., Gouda S. M., Elhaes H., Yahia I. S., Zahran H.Y. and Ibrahim M., Molecular modelling analyses for the effect of solvents on amino acids, *Biointerface Research in Applied Chemistry*. 9(5), 4379-4383 (2019).
- [6] Ezzat H. A., Hegazy M. A., Nada N. A. and Ibrahim M. A. Effect of Nano Metal Oxides on

- the Electronic Properties of Cellulose, Chitosan and Sodium Alginate, *Biointerface Research in Applied*, 9(4) 4143 - 4149 (2019).
- [7] Ibrahim A., Elhaes H., Meng F. and Ibrahim M., Effect of Hydration on the Physical Properties of Glucose, *Biointerface Research in Applied Chemistry*, 8 (4), 4114-4118 (2019).
- [8] Bayoumy A. M., Elhaes H., O. Osman, Kholmurodov K. T., Hussein T. and Ibrahim M. A., Effect of Nano Metal Oxides on Heme Molecule: Molecular and Bimolecular Approaches, *Biointerface Research in Applied Chemistry*, 10 (1), 4837-4845 (2020).
- [9] Ibrahim A., Elhaes H., Ibrahim M., Ibrahim S. Y. and Zahran H. Y., Molecular Modelling Analyses for Polyvinylidene X (X=F, Cl, Br and I), *Biointerface Research in Applied Chemistry*, 9 (2), 3890-3893, (2019).
- [10] Badry R., Shaban H., Elhaes H., Refaat A. and Ibrahim M., Molecular Modelling Analyses of Polyaniline Substituted with Alkali and Alkaline Earth Elements, *Biointerface Research in Applied Chemistry*, 8(6), 3719-3724 (2018).
- [11] Badry R., Ghanem A. S. A. E., Ahmed H., Fahmy A., Elhaes H., Refaat A. and Ibrahim M., Effect of Li, Na, K, Be, Mg and Ca on the electronic properties, geometrical parameters of carboxylic acids, *Biointerface Research in Applied Chemistry*, 8 (6), 3657-3660 (2018).
- [12] Badry R., Omar A., Mohammed H., Mohamed D. A. A., Elhaes H., Refaat A., and Ibrahim M., Effect of Alkaline Elements on the Structure and Electronic properties of Glycine, *Biointerface Research in Applied Chemistry*, 8(6), 3682-3687 (2018).
- [13] Sabry N. M., Tolba S., Abdel-Gawad F. Kh., Bassem S. M., Nassar H., El-Taweel G. E., Okasha A. and Ibrahim M., Interaction between Nano Silver and Bacteria: Modelling Approach, *Biointerface Research in Applied Chemistry*, 8(5), 3570-3574 (2018).
- [14] Sabry N. M., Tolba S., Abdel-Gawad F. Kh., Bassem S. M., Nassar H., El-Taweel G. E., Ibrahim M., On the Molecular Modelling Analyses of the Interaction between Nano Zinc Oxide and Bacteria, *Biointerface Research in Applied Chemistry*, 8 (3), 3294 - 3297 (2018).
- [15] Refaat A., Atta D., Osman O., Mahmoud A., El-Kohadary Sh., Malek W., Ferretti M., Elhaes H. and Ibrahim M., Analytical and Computational Study of Three Coptic Icons in Saint Mercurius Monastery, Egypt, *Biointerface Research in Applied Chemistry*. 9 (6), 4685-4698 (2019).
- [16] Bayoumy A. M., Elhaes H., Osman O., Hussein T. and Ibrahim M. A., Mapping Molecular Electrostatic Potential for Heme Interacting with Nano Metal Oxides, *Biointerface Research in Applied Chemistry*, 10 (2), 5091 - 5095 (2020).
- [17] Fahmy A., Khafagy R. M., Elhaes H. and Ibrahim M. A., Molecular Properties of Polyvinyl Alcohol/Sodium Alginate Composite, *Biointerface Research in Applied Chemistry*, 10 (1), 4734-4739 (2020).
- [18] Foresman J. B. and Frisch, A. Exploring chemistry with electronic structure methods, Chap., 1, computational models and model chemistry, 2nd ed., Gaussian Inc. (1996).
- [19] Hehre W. J., Radom L., Schleyer P. V. R. and Pople J. A., *Ab initio Molecular Orbital Theory*. Wiley, New York (1986).
- [20] Foresman J. B., *Ab initio Techniques in Chemistry: Interpretation and Visualization*, Chap. 14 in *What Every Chemist Should Know about Computing*, Ed. Swift, M. L. and Zielinski, T. J. Acs Books, Washington, D.C. (1996).
- [21] Elhaes H., Fakhry A. and Ibrahim M., Carbon Nano Materials as Gas Sensors, *Materials Today: Proceedings*, 3(6), 2483-2492(2016).
- [22] Kroto H. W., Heath J. R., O'Brien S. C., Curl R. F. and Smalley R. E., C<sub>60</sub>: Buckminsterfullerene, *Nature*, 318, 162-163 (1985).
- [23] Scharff P., New carbon materials for research and technology, *Carbon*, 36 (5-6), 481-486 (1998).
- [24] Hernadi K., Fonseca A., Nagy J. B., Bernaerts D., Lucas A. A., Fe-catalyzed carbon nanotube formation, *Carbon*, 34(10), 1249-1257 (1996).
- [25] Wang Z., Zhua F., Wang W. and Ruan M., Synthesis of carbon nanostructures by ion sputtering, *Physics Letter A*. 242(4-5), 261-265 (1998).
- [26] Suchanek W.L., Libera J.A., Gogotsi Y. and Yoshimura M., Behavior of C<sub>60</sub> under Hydrothermal Conditions: Transformation to Amorphous Carbon and Formation of Carbon Nanotubes, *Solid State Chemistry*, 160(1), 184-188 (2001).
- [27] Maruyama S., Miyauchi Y., Edamura T., Igarashi Y., Chiashi S. and Murakami Y., Single walled carbon nanotubes catalytically grown from mesoporous silica thin film, *Chemical Physics Letters*, 375(3-4), 553-559 (2003).
- [28] Ringor C. L. and Miyazawa K., Synthesis of C<sub>60</sub> nanotubes by liquid-liquid interfacial precipitation method: Influence of solvent ratio, growth temperature, and light illumination, *Diamond and Related Materials*, 17(4-5), 529-534 (2008).
- [29] Manafi S., Nadali H., Irani H.R., Low Temperature Synthesis of Multi-Walled Carbon Nanotubes via a Sonochemical/Hydrothermal Method, *Materials Letters*, 62(26), 4175-4176 (2008).
- [30] Yu M. F., Lourie O., Dyer M. J., Moloni K., Kelly T. F. and Ruoff R. S., Strength and breaking mechanism of multiwalled carbon nanotubes under tensile load, *Science*, 287(5453), 637-640 (2000).
- [31] Iijima S., Brabec C., Maiti A. and Bernholc J., Structural flexibility of carbon nanotubes, *Journal of Chemical Physics*, 104(5), 2089-2092 (1996).
- [32] Cobden D., Bockrath M., Chopra N., Zettle A., McEuen P., Rinzler A. and Smalley R., Spin Splitting and Even-Odd Effects in Carbon Nanotubes, *Physical Review Letters*, 81(3), 681-684 (1998).
- [33] Jung Y.J., Kar S., Talapatra S., Soldano C., Viswanathan G., Li X.S., Yao Z.L., Ou F.S., Avadhanula A., Vajtai R., Curran S., Nalamasu O. and Ajayan P.M., Aligned Carbon Nanotube-Polymer Hybrid Architectures for Flexible Electronic Applications, *Nano Letters*, 6(3), 413-418 (2006).
- [34] de Jonge N., Lamy Y., Schoots K. and Oosterkamp T.H., High brightness electron beam

- from a multi-walled carbon nanotube, *Nature*, 420(6914), 393-395 (2002).
- [35] Fennimore A. M., Yuzvinsky T. D., Han W. and Fuhrer M. S., Cumings J., Zettl A., Rotational actuators based on carbon nanotubes, *Nature*, 424(6947), 408-410 (2003).
- [36] Huang W., Taylor S., Fu K., Lin Y., Zhang D., Hanks T. W., Rao A. M. and Sun Y. P., Attaching Proteins to Carbon Nanotubes via Diimide-Activated Amidation under ambient conditions, *Nano Letters*, 2(4), 311-314 (2002).
- [37] Liu C., Fan Y.Y., Liu M., Cong H.T., Cheng H.M. and Dresselhaus M.S., Hydrogen Storage in Single-Walled Carbon Nanotubes at Room Temperature, *Science*, 286(5442), 1127-1129 (1999).
- [38] Villers D., Sun S.H., Serventi A.M., Dodelet J.P. and De'silets S., Characterization of Pt Nanoparticles Deposited onto Carbon Nanotubes Grown on Carbon Paper and Evaluation of This Electrode for the Reduction of Oxygen, *Journal of Physical Chemistry B*, 110 (51), 25916-25925 (2006).
- [39] Castro Neto A. H., Guinea F., Peres N. M. R., Novoselov K. S., and Geim A. K., The electronic properties of graphene, *Reviews of Modern Physics*, 81(1),109 (2009)
- [40] Chen J.-H., Jang C., Adam S., Fuhrer M. S., Williams E. D. and Ishigami M., Charged Impurity Scattering in Graphene, *Nature Physics*, 4, 377-381 (2008).
- [41] Pi K., McCreary K. M., Bao W., Han W., Chiang Y. F., Li Y., Tsai S.-W., Lau C. N. and Kawakami R. K., Electronic doping and scattering by transition metals on graphene, *Physical Review B*, 80(6), 075406 (2009).
- [42] Wang Y., Bao W., Xiao S., Fuhrer M. S., and Reutt-Robey J., Electrical detection of  $\text{CF}_3\text{Cl}$  phase transitions on graphene, *Applied Physics Letters*,103, 201606 (2013).
- [43] Subrahmanyam K.S., Manna A. K., Pati S. K. and Rao C.N.R., A study of graphene decorated with metal nanoparticles, *Chemical Physics Letters*, 497(1-3), 70-75 (2010).
- [44] Zhang X., Yu L., Wu X., Hu W., Experimental Sensing and Density Functional Theory Study of  $\text{H}_2\text{S}$  and  $\text{SO}_2$  Adsorption on Au-Modified Graphene, *Advanced Science*, 2(11), 1500101, (2015).
- [45] Gadipelli S. and Guo Z.X., Graphene-based materials: Synthesis and gas sorption, storage and separation, *Progress in Materials Science*, 69, 1-60, (2015).
- [46] Novoselov K.S., Falko V.I., Colombo L., Gellert P.R., Schwab M.G., Kim K., A roadmap for graphene, *Nature*, 490, 192-200, (2012).
- [47] Wang X., Gao D., Li M., Li H., Li C., Wu X., Yang B., CVD graphene as an electrochemical sensing platform for simultaneous detection of biomolecules, *Scientific Reports*, 7, 7044, (2017).
- [48] Haddon R. C., Schneemeyer L. F., Waszczak J. V., Glarum S. H., Tycko R.,Dabbagh G., Kortan A. R., Muller A. J., Mujsce A. M., Rosseinsky M. J., Zahurak S. M., Makhija A. V., Thiel F. A., Raghavachari K., Cockayne E. and Elser V., Experimental and theoretical determination of the magnetic susceptibility of  $\text{C}_{60}$  and  $\text{C}_{70}$ , *Nature*, 350, 46-47 (1991).
- [49] Aryasetiawan F., Gunnarsson O., Koch E. and Martin R.M., Pauli susceptibility of  $\text{A}_3\text{C}_{60}$ (A=K,Rb), *Physical Review B*, 55(16), R10165 (1997).
- [50] Elserlser V. and Haddon R. C., Icothedral  $\text{C}_{60}$  – an aromatic molecule with a vanishingly small ring current magnetic susceptibility, *Nature*, 325, 792-794 (2002).
- [51] Andreoni W., Small semiconductor clusters and fullerenes: Structural and electronic properties from ab-initio molecular dynamics, *Nanostructured Materials*, 3, 293-300(1993).
- [52] Raghavachari K., Sosa C., Fullerene derivatives. Comparative theoretical study of  $\text{C}_{60}\text{O}$  and  $\text{C}_{60}\text{CH}_2$ , *Chemical Physics Letters*, 209(3), 223-228 (1993).
- [53] Suzuki S., Torisu H., Kubota H., Wakabayashi T., Shiromaru H. and Achiba Y., Formation and stability of small metallocarbon clusters: what is the specificity for the formation of stable metallofullerenes? , *International Journal of Mass Spectrometry, Ion Proc.* 138, 297-306 (1994).
- [54] Scuseria G. E., The equilibrium structures of giant fullerenes: faceted or spherical shape? An ab initio Hartree-Fock study of icosahedral  $\text{C}_{240}$  and  $\text{C}_{540}$  ,*Chemical Physics Letters*, 243, 193-198 (1995).
- [55] Zandler M. E., Behrman E. C., Arrasmith M. B., Myers J. R. and Smith T. V., Semiempirical molecular orbital calculation of geometric, electronic, and vibrational structures of metal oxide, metal sulfide, and other inorganic fullerene spheroids, *Journal of Molecular Structure: THEOCHEM*, 362(2), 215-224 (1996).
- [56] Choho K., Van de Woude G., Van Lier G. and Geerlings P., An ab initio quantum chemical study on the structure, stability and polymerization of  $\text{C}_{28}$  and its derivatives, *Journal of Molecular Structure: THEOCHEM*, 417(3), 265-276 (1997).
- [57] Sánchez-Portal D., Artacho E., Pascual J. I., Gómez-Herrero J. and Martin R. M., First principles study of the adsorption of  $\text{C}_{60}$  on Si(1 1 1), *Surface Science*, 482-485, 39-43 (2001).
- [58] Godwin P. D., Kenny S. D. and Smith R., The bonding sites and structure of  $\text{C}_{60}$  on the Si(100) surface, *Surface Science*, 529(1-2), 237-246 (2003).
- [59] Chang J. G., Hwang C. C., Ju S. P. and Huang S. H., A molecular dynamics simulation investigation into the structure of fullerene  $\text{C}_{60}$  grown on a diamond substrate, *Carbon*, 42(12-13), 2609-2616 (2004).
- [60] Gayathri S. S. and Patnaik A., A new fullerene  $\text{C}_{60}$ -didodecyloxy benzene dyad: An evidence for ground state electron transfer, *Chemical Physics Letters*, 414(1-3), 198-203 (2005).
- [61] Ibrahim M. and El-Haes H., Spectroscopic study of  $\text{C}_{60}$  and  $\text{C}_{80}$  and their epoxides, *Chinese Journal of Physics*, 43(5), 915-923 (2006).
- [62] Hebard A. F., Rosseinsky M. J., Haddon R. C., Murphy D. W., Glarum S. H., Palstra T. T. M., Ramirez A. P., and Kortan A. R., Superconductivity at 18 K in potassium-doped  $\text{C}_{60}$ , *Nature*, 350 (6319), 600-601 (1991).
- [63] Holczer K., Klein O., Gruner G., Thompson J. D., Diederich F. and Whetten R. L., Critical magnetic fields in the superconducting state of

- $K_3C_{60}$ , Physical Review Letters, 67(2), 271274 (1991).
- [64] Haddon R. C., Electronic structure, conductivity and superconductivity of alkali metal doped  $C_{60}$ , Pure and Applied Chemistry, 65(1), 11-15 (1993).
- [65] Giannozzi P. and Andreoni W., Effects of Doping on the Vibrational Properties of  $C_{60}$  from First Principles:  $K_6C_{60}$ , Physical Review Letters, 76(26), 4915-4918 (1996).
- [66] Schoen J. H., Kloc Ch. and Batlogg B., Superconductivity at 52 K in hole-doped  $C_{60}$ , Nature, 408, 549-552 (2000).
- [67] Schoen J. H., Kloc Ch. and Batlogg B., High-Temperature Superconductivity in Lattice-Expanded  $C_{60}$ , Science, 293(5539), 2432-2434 (2001).
- [68] Yang C. H., Qiao J., Sun Q. J., Jiang K. J., Li Y. L. and Li Y. F., Improvement of the performance of polymer/ $C_{60}$  photovoltaic cells by small-molecule doping, Synthetic Metals, 137(1-3), 1521-1522 (2003).
- [69] Goto T. and Maezawa M., Fabrication of filamentary potassium-doped  $C_{60}$  superconductors by suspension spinning method, Physica C: Superconductivity, 614(1), 412-414, (2004).
- [70] Lee K. and Kim H., Polymer photovoltaic cells based on conjugated polymer-fullerene composites, Current Applied Physics, 4(2-4), 323-326 (2004).
- [71] Majumdar H. S., Baral J. K., Oesterbacka R., Ikkala O. and Stubb H., Fullerene-based bistable devices and associated negative differential resistance effect, Organic Electronics, 6(4), 188-192 (2005).
- [72] Artru X., Fomin S.P., Shul'ga N.F., Ispirian K.A. and Zhevago N.K., Carbon nanotubes and fullerenes in high-energy and X-ray physics, Physics Reports, 412(2-3), 89-189 (2005).
- [73] Zhang X.-X., Liu Tang W.-T. and J., Xiao P., Study on PD detection in  $SF_6$  using multiwall carbon nanotube films sensor, IEEE Trans. Dielectric and Electrical. Insulation, 17(3), 838-844 (2010).
- [74] Kong J., Franklin N.R., Zhou C., Chapline M.G., Peng S., Kyeongjae C. and Hongjie D., Nanotube molecular wires as chemical sensors, Science, 287(5453), 622-625 (2000).
- [75] Zhang X.-X., Bing Y., Wang X.-J. and Luo C.-C., Effect of plasma treatment on multiwalled carbon nanotubes for the detection of  $H_2S$  and  $SO_2$ , Sensors, 12(7), 9375-9385 (2012).
- [76] Zhang X.-X., Bing Y., Dai Z.-Q. and Luo C.-C., The gas response of hydroxyl modified SWCNTs and carboxyl modified SWCNTs to  $H_2S$  and  $SO_2$ , PrzegladElektrotechniczny, 88(7), 311-314 (2012).
- [77] Srivastava R., Suman H., Shrivastava S. and Srivastava A., DFT Analysis of Pristine and functionalized Zigzag CNT: A case of  $H_2S$  sensing, Chemical Physics Letters, 731(16), 136575 (2019).
- [78] Liu Y., Peng Y., An B., Li L. and Liu Y., Effect of molecular structure on the adsorption affinity of sulfonamides onto CNTs: Batch experiments and DFT calculations, Chemosphere, 246, 125778 (2020).
- [79] Wei-Cheng T., Hung-Ling L., Chun-Yen K., Chia-Jung L., Chang-Jung H., Sensors, IEEE 201, 1036-1039 (2011).
- [80] Hizhnyi Y., Nedilko S.G., Borysiuk V. and Gubanov V.A., Computational studies of boron- and nitrogen-doped single-walled carbon nanotubes as potential sensor materials of hydrogen halide molecules HX (X=F, Cl, Br), International Journal of Quantum Chemistry, 115(20), 1475-1482 (2015).
- [81] Fam D.W.H., Palaniappan A.L., Tok A.I.Y., Liedberg B. and Mochhala S.M., A review on technological aspects influencing commercialization of carbon nanotube sensors, Sensors and Actuators B: Chemical, 157(1), 1-7 (2011).
- [82] Mittal M. and Kumar A., Carbon nanotube (CNT) gas sensors for emissions from fossil fuel burning, Sensors and Actuators B: Chemical, 203, 349-362 (2014).
- [83] Lee K., Scardaci V., Kim H.-Y., Hallam T., Nolan H., Bolf B.E., Maltbie G.S., Abbott J.E. and Duesberg G.S., Highly sensitive, transparent, and flexible gas sensors based on gold nanoparticle decorated carbon nanotubes, Sensors and Actuators B: Chemical, 188, 571-575 (2013).
- [84] Mendoza F., Hernandez D.M., Makarov V., Febus E., Weiner B.R. and Morell G., Room temperature gas sensor based on tin dioxide-carbon nanotubes composite films, Sensors and Actuators B: Chemical, 190, 227-233 (2014).
- [85] Adjizian J.-J., Leghrib R., Koos A.A., Suarez-Martinez I., Crossley A., Wagner P., Grobert N., Llobet E. and Ewels C.P., Boron- and nitrogen-doped multi-wall carbon nanotubes for gas detection, Carbon, 66, 662-673 (2014).
- [86] Bai L. and Zhou Z., Computational study of B- or N-doped single-walled carbon nanotubes as  $NH_3$  and  $NO_2$  sensors, Carbon, 45(10), 2105-2110 (2007).
- [87] Wang Y., and Yeow J.T.W., A Review of Carbon Nanotubes-Based Gas Sensors, Journal of Sensors, 2009 Article ID 493904, 1-24 (2009).
- [88] Pavageau M.P., PeaCheyran Ch., Krupp E. M., Morin A. and Donard O. F.X., Volatile Metal Species in Coal Combustion Flue Gas, Environmental Science and Technology, 36(7), 1561-1573 (2002).
- [89] Mollania F., Hadipour N. L., Mollania N., CNT-based nanocarrier loaded with pyrimethamine for adipose mesenchymal stem cells differentiation and cancer treatment: The computational and experimental methods, Journal of Biotechnology, 308, 40-55 (2020).
- [90] Saha B. and Bhattacharyya P. Kr., Anion- $\pi$  interaction in oxoanion-graphene complex using coronene as model system: A DFT study, Computational and Theoretical Chemistry, 1147, 62-71(2019).
- [91] Mohammadi-rad N., Esrafil M. D. and Sardroodi J. J.,  $CuN_3$  doped graphene as an active electrocatalyst for oxygen reduction reaction in fuel cells: A DFT study, Journal of Molecular Graphics and Modelling, 96, 107537 (2020).
- [92] Guo B., Fang L., Zhang B. and Gong J.R., Graphene doping: a review, Insciences Journal, 1(2), 80-89 (2011).

- [93] Tsetseris L., Wang B. and Pantelides S., Substitutional doping of graphene: the role of carbon divacancies, *Physical Review B*, 89(3), 035411(2014).
- [94] Gholizadeh R., Yu Y.-X.,  $N_2O^+$  CO reaction over Si-and Se-doped graphenes: an ab initio DFT study, *Applied Surface Science*, 357, 1187-1195 (2015).
- [95] Novoselov K.S., Geim A.K., Morozov S.V., Jiang D., Katsnelson M.I., Grigorieva I.V., Dubonos S.V. and Firsov A.A., Two-dimensional gas of massless Dirac fermions in graphene, *Nature*, 438, 197-200 (2005).
- [96] Zhang Y., Tan Y.-W., Stormer H.L., Kim P., Experimental observation of the quantum Hall effect and Berry's phase in graphene, *Nature*, 438, 201-204(2005).
- [97] Geim A.K. and Novoselov K.S., The rise of graphene, *Nature Materials*, 6, 183-191 (2007).
- [98] Danneau R., Wu F., Craciun M.F., Russo S., Tomi M.Y., Salmilehto J., Morpurgo A.F., Hakonen P.J., Shot Noise in Ballistic Graphene, *Physical Review Letters*, 100(19), 196802 (2008).
- [99] Schedin F., Geim A.K., Morozov S.V., Hill E.W., Blake P., Katsnelson M.I. and Novoselov K.S., Detection of individual gas molecules adsorbed on graphene, *Nature Materials*, 6, 652-655 (2007).
- [100] Varghese S. S., Swaminathan S., Singh K. K. and Mittal V., Ab initio study on gas sensing properties of group III (B, Al and Ga) doped graphene, *Computational Condensed Matter*, 9, 40-55 (2016).
- [101] Shteplyuk I., Caffrey N. M., Lakimov T., Khranovskyy V., Abrikosov I. A. and Yakimova R., On the interaction of toxic Heavy Metals (Cd, Hg, Pb) with graphene quantum dots and infinite grapheme, *Scientific Reports*, 7, 3934 (2017).
- [102] Shteplyuk I., Eriksson J., Khranovskyy V., Iakimov T., Lloyd Spetz A. and Yakimova R., Monolayer graphene/SiC Schottky barrier diodes with improved barrier height uniformity as a sensing platform for the detection of heavy metals, *Beilstein Journal Nanotechnology*, 7, 1800-1814 (2016).
- [103] Darvishnejad M. H. and Reisi-Vanani A., Multiple CO<sub>2</sub> capture in pristine and Sr-decorated graphyne: A DFT-D3 and AIMD study, *Computational Materials Science*, 176, 109539 (2020).
- [104] Sukhbir Singh and Inderpreet Kaur, Bandgap engineering in armchair graphene nanoribbon of zigzag-armchair-zigzag based Nano-FET: A DFT investigation, *Physica E*, 118, 113960 (2020).
- [105] Mofidi F. and Reisi-Vanani A., Investigation of the electronic and structural properties of graphene oxide toward CO, CO<sub>2</sub> and NH<sub>3</sub> adsorption: A DFT and MD study, *Applied Surface Science*, 507, 145134 (2020).
- [106] Nemati-Kande E., Karimian R., Goodarzi V. and Ghazizadeh E., Feasibility of pristine, Al-doped and Ga-doped Boron Nitride nanotubes for detecting SF<sub>4</sub> gas: A DFT, NBO and QTAIM investigation, *Applied Surface Scienc*, 510, 145490 (2020).
- [107] Ezzat H., Badry R., Yahia I.S., Zahran H.Y., Ibrahim A., Elhaes H. and Ibrahim M.A., Mapping the molecular electrostatic potential of fullerene, *Egyptian Journal of Chemistry*, 62(6), 1391-1402 (2019).
- [108] Chung D.D.L., Electromagnetic interference shielding effectiveness of carbon materials, *Carbon*, 39(2), 279-285 (2001).
- [109] Luo X, Chugh R, Biller B.C, Hoi Y.M, Chung D.D.L., Electronic Applications of Flexible Graphite, *Journal of Electronic Materials*, 315, 535-544 (2002).
- [110] Celzard A., Marêché J.F., Furdin G., Modelling of exfoliated graphite, *Progress in Materials Science*, 50 (1), 93-179 (2005).
- [111] Malifarge S., Delobel B. and Delacourt Ch., Experimental and Modelling Analysis of Graphite Electrodes with Various Thicknesses and Porosities for High-Energy-Density Li-Ion Batteries, *Journal of The Electrochemical Society*, 165(7), A1275-A1287 (2018).
- [112] Kyani A. and Diudea M. V., Molecular dynamics simulation study of the diamond D5 substructures, *Central European Journal of Chemistry*, 10(4), 1028-1033 (2012).
- [113] Huang Ch., Peng X., Yang B., Zhao Y., Xiang H., Chen X., Li Q. and Fu T., Molecular dynamics simulations for responses of nanotwinned diamond films under nanoindentation, *Ceramics International*, 43(18), 16888-16894 (2017).
- [114] Awan I. S., Xiaoqun W., Pengcheng H. and Shanyi D., Developing an approach to calculate carbon fiber surface energy using molecular simulation and its application to real carbon fibers, *Journal of Composite Materials*, 46(6), 707-715 (2011).
- [115] Desai S., Li Ch., Shen T. and Strachan Al., Molecular modelling of the microstructure evolution during carbon fiber processing, *Journal of Chemical Physics*, 147(22), 224705 (2017).
- [116] Sharma K., Shukla M., Molecular modelling of the mechanical behavior of carbon fiber-amine functionalized multiwall carbon nanotube/epoxy composites, *New Carbon Materials*, 29 (2), 132-142 (2014).
- [117] Ezzat H., Badry R., Yahia I.S., Zahran H.Y., Elhaes H. and Ibrahim M.A., Mapping the molecular electrostatic potential of carbon nanotubes, *Biointerface Research in Applied Chemistry*, 8(5), 3539-3542 (2018).
- [118] Ibrahim M. A., Elhaes H., El-Khodary S. A., Morsy M., Refaat A., Yahia I. S. and Zahran H. Y., Molecular Modelling Analyses for the Effect of Alkali Metal Oxides on Graphene, *Biointerface Research in Applied Chemistry*, 8(5), 3522-3525 (2018).
- [119] Bashamavaz H., Habibi-Yangjeh A., Kamali S.H., A first-principle investigation of NO<sub>2</sub> adsorption behavior on Co, Rh, and Ir-embedded graphitic carbon nitride: looking for highly sensitive gas sensor, *Physics Letters A*, 384(2), 126057 (2020).
- [120] Ghashghae M., Ghambarian M., Azizi Z., Chemistry of black phosphorus, in: R. Boddula Inamuddin, A.M. Asiri (Eds.), *Black Phosphorus: Synthesis, Properties and Applications*, Springer International Publishing, Cham, 59-72(2020).
- [121] Ghashghae M., Azizi Z., Ghambarian M., Conductivity tuning of charged triazine and

- heptazine graphitic carbon nitride (g-C<sub>3</sub>N<sub>4</sub>) quantum dots via nonmetal (B, O, S, P) doping: DFT calculations, *Journal of Physics and Chemistry of Solids*, 141, 109422 (2020).
- [122] Becke A.D., Density-functional thermochemistry. III. The role of exact exchange. *Journal of Chemical Physics*, 98(7), 5648–5652(1993).
- [123] Lee C., Yang W. and Parr R.G., Development of the Colle-Salvetti correlation-energy formula into a functional of the electron density, *Physical Review B*, 37(2), 785 (1988).
- [124] Vosko S.H., Wilk L., Nusair M., Accurate spin-dependent electron liquid correlation energies for local spin density calculations: a critical analysis, *Canadian Journal of Physics*, 58(8), 1200-1211(1980).
- [125] Nematov D.D., Burhonzoda A.S., Khuseinov M.A., Kholmurodov K.T., Elhaes H., and Ibrahim M. A., The Quantum-Chemistry Calculations of Electronic Structure of Boron Nitride Nanocrystals with Density Functional Theory Realization, *Egyptian Journal of Chemistry. Special issue, the First International Conference on Molecular Modelling and Spectroscopy*, 21-27, (2019).
- [126] El Gabaly S.G., Youssif G. M., Bayoumy A. M., Ezzat H., Elhaes H., Refaat A., and Ibrahim M. A., Modelling The Effect of Functional Groups on The Electronic Properties of Benzene, Pyridine and Pyrimidine, *Egyptian Journal of Chemistry, Special issue, the First International Conference on Molecular Modelling and Spectroscopy*, 1-11, (2019).
- [127] Doroshkevich A.S., Nabiev A.A., Shylo A.V., Pawlukojć A., Doroshkevich V. S., Glazunova V.A., Zelenyak T. Y., Doroshkevich N.V., Rahmonov K. R., Khamzin E. Kh., Nematov D. D., Burhonzoda A. S., Khuseinov M. A., Kholmurodov K. T., Majumder S., Balasoju M., Madadzada A., Bodnarchuk V. I., and Ibrahim M. A., “Frequency Modulation of The Raman Spectrum at The Interface DNA - ZrO<sub>2</sub> Nanoparticles”, *Egyptian Journal of Chemistry. Special issue, the First International Conference on Molecular Modelling and Spectroscopy*, 13-20m (2019).
- [128] Mehrani S., Tayyari S. F.k., Heravi M.M., Morsali A., Vibrational spectra, normal coordinate analysis, and structure of keto form of acetylacetone. A DFT approach, *Egyptian Journal of Chemistry*. 63 (4), 10-11(2020)
- [129] Ghasemi A., Fazaeli R., Ghiasi R., DFT and NBO Studies of Stability, Electronic, and Structural Features of the 2-fluoroacetaldehyde Conformers, *Egyptian Journal of Chemistry*. 63(1), 315-324 (2020)
- [130] Nawar F., Al-Asadi R., and Abid D., Synthesis, Antibacterial Activity and DFT Calculations of Some Thiazolidine-4-Carboxylic acid Derivatives and Their Complexes with Cu(II), Fe(II) and VO(II), *Egyptian Journal of Chemistry*, 63(1), 349-362 (2020).